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Supporting Information

for

Structure-Nucleophilicity Relationships for Enamines

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A. Products of the reactions of enamines, pyrroles, or indoles with benzhydrylium salts

The synthesis of enamines 1a-q, and their reactions with benzhydrylium salts $Ar_2CH^+ BF_4^$ were performed under exclusion of moisture in an atmosphere of dry nitrogen in carefully dried Schlenk glassware. Dichloromethane was freshly distilled from CaH₂ before use.

<u>Procedure A:</u> A solution of the freshly distilled or recrystallized enamine was added dropwise to a stirred solution of the benzhydrylium salt in CH_2Cl_2 at room temperature. After fading of the color (in cases of reversible reactions after 2 h at the latest) the solvent was evaporated in vacuo to yield the crude product, which was washed with dry Et_2O and dried several hours in vacuo (10^{-2} mbar).

<u>Procedure B:</u> The crude product obtained by Procedure A was dissolved in dilute HCl and stirred for 30 min. The solution was then neutralized by treatment with dilute NaOH. Extraction with CH_2Cl_2 (3 × 30 mL), drying of the combined organic layers with MgSO₄, filtration, and evaporation of the solvent in vacuo gave a product which was purified by recrystallization.

<u>Procedure C:</u> A solution of the benzhydryl salt in dichloromethane (25 mL) was added dropwise to a stirred solution of 10 equiv. of the freshly distilled or recrystallized pyrrole or indole in dichloromethane (25 mL) (An excess of the nucleophile is necessary for trapping the protons which are released during the electrophilic substitution. Reactions with equimolar amounts of the reactants usually do not proceed quantitatively and lead to the formation of side products, e.g., disubstituted arenes). After fading of the color, the solvent was evaporated in vacuo to yield the crude product, which was purified by column chromatography.

N-(2-(Bis(julolidin-9-yl)methyl)cyclopentylidene)pyrrolidinium tetrafluoroborate (2a) was obtained from 1-(*N*-pyrrolidino)cyclopentene (1a) (55 mg, 0.40 mmol) and (jul)₂CH⁺ BF₄⁻ (0.18 g, 0.40 mmol) as an orange solid (144 mg, 62 %) following Procedure A. ¹H NMR (300 MHz, CDCl₃): = 1.50–2.25 (m, 16 H), 2.52–3.00 (m, 10 H), 3.01–3.15 (m, 8 H), 3.48–4.05 (m, 6 H), 6.42, 6.64 (2 s, 2×2 H, ArH); ¹³C NMR (75.5 MHz, CDCl₃): = 20.2 (t, C-4), 22.0 (t, CH₂), 24.6/24.8 (2 t, CH₂), 27.7 (t, CH₂), 30.0, 34.4 (2 t, C-3 and C-5), 50.5 (t, CH₂), 50.9, 53.7 (2 d, Ar₂CH and C-2), 55.0/55.2 (2 t, CH₂), 121.7 (s, Ar), 126.6/126.9 (2 d, Ar), 128.7 (s, Ar), 141.9/142.1 (2 s, Ar), 197.0 (s, C-1).



2-(Bis(*N***-methyl-1,2,3,4-tetrahydroquinolin-6-yl)methyl)cyclopentanone** (3a) was obtained from 1-(*N*-pyrrolidino)cyclopentene (1a) (137 mg, 1.00 mmol) and (thq)₂CH⁺ BF₄⁻ (392 mg, 1.00 mmol) following Procedure B. Crystallization (EtOH) of the crude product gave a brown solid (207 mg, 39 %). ¹H NMR (300 MHz, CDCl₃): = 1.65–2.29 (m, 10 H, 5 × CH₂), 2.59–2.86 (m, 11 H, 2 × CH₂, 2 × NMe, 2-H), 3.13–3.18 (m, 4 H, 2 × CH₂), 4.40 (d, ³*J* = 4.4 Hz, 1 H, Ar₂C*H*), 6.42–7.00 (m, 6 H, ArH); ¹³C NMR (75.5 MHz, CDCl₃): = 20.6 (t, C-4), 22.5/22.6 (2 t, CH₂), 27.1 (t, C-3), 27.8/27.9 (2 t, CH₂), 38.6 (t, C-5), 39.1/39.2 (2 q, NMe), 48.2 (d, Ar₂CH), 51.30/51.35 (2 t, CH₂), 53.7 (d, C-2), 110.6/110.8 (2 d, Ar), 122.4/122.7 (2 s, Ar), 126.5/127.7, 129.1/129.6 (2 × 2 d, Ar), 130.9/132.0 (2 s, Ar), 145.0 (s, Ar), 220.4 (s, C-1); MS (70 eV, EI): *m/z* (%): 388 (4) [M⁺], 305 (100) [Ar₂CH⁺], 269 (52), 207 (12).



2-(Bis(4-dimethylaminophenyl)methyl)cyclohexanone¹⁷ (**3b**) was obtained from 1-(*N*-pyrrolidino)cyclohexene (**1b**) (151 mg, 0.998 mmol) and $(dma)_2CH^+$ BF₄⁻ (340 mg, 1.00 mmol) in CH₂Cl₂ (50 mL) following Procedure B. Recrystallization (EtOH) furnished **3b** as a light-brown solid (229 mg, 65 %) which showed identical ¹H and ¹³C NMR spectra as the

corresponding sample that was obtained from 1-(morpholino)cyclohexene and $(dma)_2CH^+$ BF₄⁻ in ref. 17.

N-(2-(Bis(julolidin-9-yl)methyl)cyclopentylidene)piperidinium tetrafluoroborate (2c) was obtained from 1-(*N*-piperidino)cyclopentene (1c) (54 mg, 0.36 mmol) and (jul)₂CH⁺ BF₄⁻ (160 mg, 0.360 mmol) as a dark orange solid (190 mg, 89 %) following Procedure A. ¹H NMR (300 MHz, CDCl₃): = 1.30–2.22 (m, 18 H), 2.45–2.87 (m, 8 H), 2.90–3.11 (m, 10 H), 3.27–3.90 (m, 3 H), 4.22 (d, ³*J* = 4.4 Hz, 1 H, Ar₂C*H*), 6.41, 6.55 (2 s, 2 × 2 H, ArH); MS (FAB): m/z (%): 504 (1), 464 (2), 442 (12), 438 (10), 373 (4), 357 (100) [Ar₂CH⁺].



2-(Bis(4-dimethylaminophenyl)methyl)cyclopentanone⁴³ (**3c**) was obtained from 1-(*N*-piperidino)cyclopentene (**1c**) (153 mg, 1.01 mmol) and $(dma)_2CH^+BF_4^-$ (340 mg, 1.00 mmol) following Procedure B. Crystallization (EtOH) of the crude product gave a light brown solid (166 mg, 49 %) which showed identical ¹H and ¹³C NMR spectra as the corresponding sample that was obtained from 1-(trimethylsiloxy)cyclopentene and $(dma)_2CH^+BF_4^-$ in ref. 43.

2-(Bis(lilolidin-8-yl)methyl)cyclopentanone (**3e**) was obtained from 1-(*N*-morpholino)-cyclopentene (**1e**) (0.216 mL, 1.35 mmol) and (lil)₂CH⁺ BF₄⁻ (510 mg, 1.23 mmol) at -90 °C as described in Procedure B. ¹H NMR (300 MHz, CDCl₃): = 1.66–1.79 (m, 3 H, 3-H and 4-H₂), 1.99–2.10 (m, 4 H, 2 × CH₂), 2.15–2.27 (m, 3 H, 3-H and 5-H₂), 2.56–2.67 (m, 4 H, 2 × CH₂), 2.72–2.95 (m, 9 H, 2-H and 4 × CH₂), 3.12–3.23 (m, 4 H, 2 × CH₂), 4.42 (d, ³*J* = 4.3 Hz, 1 H, Ar₂CH), 6.51, 6.62, 6.73, 6.85 (4 s, 4 × 1 H, ArH); ¹³C NMR (75.5 MHz, CDCl₃): = 20.43 (t, C-4), 23.19/23.25, 23.82, 23.87 (2 × 2 t, CH₂), 27.22 (t, C-3), 28.67/28.76 (2 t, CH₂), 38.37 (t, C-5), 47.48/47.56 (2 t, CH₂), 49.49 (d, Ar₂CH), 53.81 (d, C-2), 55.20/55.29 (2 t, CH₂), 118.65/118.78 (2 s, Ar), 121.96/122.58, 125.82/126.75 (2 × 2 d, Ar), 128.28/128.55, 133.76/134.90 (2 × 2 s, Ar), 148.05 (s, Ar), 220.08 (s, C-1), signal assignments are based on ¹H, ¹³C-COSY experiments; MS (EI, 70 eV): m/z (%): 412 (14) [M⁺], 329 (100), 165 (12).



2-(Bis(4-dimethylaminophenyl)methyl)cyclopentenylidene)morpholinium tetrafluoroborate (2e) was obtained from 1-(*N*-morpholino)cyclopentene (1e) (0.168 mL, 1.12 mmol) and (dma)₂CH⁺ BF₄⁻ (380 mg, 1.12 mmol) following Procedure A: pale green powder (498 mg, 90 %). ¹H NMR (300 MHz, CDCl₃): = 1.89–2.35 (m, 4 H, 3-H and 4-H), 2.70–2.79 (m, 1 H, $\frac{1}{2} \times CH_2$), 2.86–2.95 (m, partially superimposed, 1 H, $\frac{1}{2} \times CH_2$), 2.86, 2.90 (2 s, 2 × 6 H, NMe₂), 3.00–3.29 (m, 2 H, 5-H₂), 3.60–4.04 (m, 8 H, 2-H, Ar₂CH, 3 × CH₂), 6.59–6.69 (m, 4 H, ArH), 7.02–7.05, 7.15–7.18 (2 m, 2 × 2 H, ArH); ¹³C NMR (75.5 MHz, CDCl₃): = 19.85 (t, C-4), 29.57 (t, C-3), 33.13 (t, C-5), 40.34/40.39 (2 q, NMe₂), 50.74, 51.09 (2 d, Ar₂CH and C-2), 54.23/54.89, 64.69/65.10 (2 × 2 t, CH₂), 112.68/112.77 (2 d, Ar), 127.74/128.69 (2 s, Ar), 128.91/129.40 (2 d, Ar), 149.52/149.88 (2 s, Ar), 200.32 (s, C-1), signal assignments are based on ¹H, ¹H- and ¹H, ¹³C-COSY experiments.



3-(Bis(4-dimethylaminophenyl)methyl)-1-methylpiperidin-4-one (3g) was obtained from 1,2,5,6-tetrahydro-1-methyl-4-(*N*-morpholino)pyridine (**1g**) (150 mg, 0.823 mmol) and (dma)₂CH⁺ CF₃SO₃⁻ (231 mg, 0.574 mmol) following Procedure B. Crystallization of the crude product gave **3g** (160 mg, 76 %) as colorless needles. M.p. 162–163 °C (decomp., from ethanol); ¹H NMR (300 MHz, CDCl₃): = 2.28 (s, 1-Me), 2.32–2.45 (m, 1 H, 5-H), 2.47–2.50 (m, 2 H, 2-H₂), 2.52–2.68 (m, 2 H, 5-H and 6-H), 2.76–2.83 (m, 1 H, 6-H), 2.86, 2.89 (2s, NMe₂), 3.25–3.31 (m, 1 H, 3-H), 4.45 (d, *J* = 11.2 Hz, 1 H, Ar₂C*H*), 6.60–6.67, 7.10–7.19 (2 m, 2 × 4 H, ArH); ¹³C NMR (75.5 MHz, CDCl₃): = 40.64 (q, NMe₂), 40.83 (t, C-5), 45.29 (q, 1-Me), 48.37 (d, Ar₂CH), 55.33 (d, C-3), 57.15 (t, C-6), 60.09 (t, C-2), 112.74/112.85, 128.40/128.67 (2 × 2 d, Ar), 131.26/131.41, 148.94/149.05 (2 × 2 s, Ar), 211.00 (s, C-4), signal assignments are based on ¹H, ¹H- and ¹H, ¹³C-COSY-experiments; MS (EI, 70 eV): *m/z* (%): 365 (8) [M⁺], 254 (18), 253 (100); elemental analysis calcd (%) for C₂₃H₃₁N₃O (365.52): C 75.58, H 8.55, N 11.50; found: C 75.55, H 8.58, N 11.56.



N-(3,3-Bis(4-(phenyl(2,2,2-trifluoroethyl)amino)phenyl)-2,2-dimethylpropylidene)morpholinium tetrafluoroborate (2j) was generated by mixing (*N*-morpholino)isobutene (1j) (22 mg, 0.16 mmol) and (pfa)₂CH⁺ BF₄⁻ (92 mg, 0.15 mmol) in CDCl₃ (1 mL). The reaction mixture was transferred into an NMR tube and analyzed without isolation. ¹H NMR (200 MHz, CDCl₃): = 1.50 (s, 6 H, CH₃), 3.35, 3.67, 3.90, 4.00 (4 m_c, 4 × 2 H, morpholino-CH₂), 4.25 (q, *J*(H,F) = 8.7 Hz, 4 H, NC*H*₂CF₃), 4.40 (s, 1 H, Ar₂C*H*), 6.82–7.35 (m, 18 H, ArH), 8.52 (s, 1 H, N⁺=CH).



3,3-Bis(4-dimethylaminophenyl)-2,2-dimethylpropanal (**3j**) was obtained from (*N*-morpholino)isobutene (**1j**) (270 mg, 1.91 mmol) and $(dma)_2CH^+ CF_3SO_3^-$ (379 mg, 0.942 mmol) following Procedure B. Crystallization from ethanol gave **3j** as beige needles (170 mg, 52 %). M.p. 102 °C (ethanol); ¹H NMR (300 MHz, CDCl₃): = 1.12 (s, 6 H, 2-CH₃), 2.89 (s, 12 H, NMe₂), 3.99 (s, 1 H, Ar₂CH), 6.60–6.66, 7.08–7.13 (2 m, 2 × 4 H, ArH), 9.69 (s, 1 H, 1-H); ¹³C NMR (75.5 MHz, CDCl₃): = 21.5 (q, 2-CH₃), 40.5 (q, NMe₂), 49.8 (s, C-2), 56.9 (d, Ar₂CH), 112.3 (d, Ar), 129.3 (s, Ar), 130.2 (d, Ar), 149.0 (s, Ar), 207.0 (d, C-1); MS (EI, 70 eV): *m/z* (%): 324 (2) [M⁺], 254 (34), 253 (100), 237 (14), 126 (20); elemental analysis calcd (%) for C₂₁H₂₈N₂O (324.47): C 77.74, H 8.70, N 8.63; found: C 77.74, H 8.87, N 8.73.



3,3-Bis(4-diphenylaminophenyl)-2-phenylpropanal (**3k**) was obtained from (*E*)- -(*N*-morpholino)styrene (**1k**) (189 mg, 1.00 mmol) and $(dpa)_2CH^+$ BF₄⁻ (589 mg, 1.00 mmol)

following Procedure B. Crystallization of the crude product gave a pale green solid (432 mg, 70 %). M.p. 130–131 °C (CH₂Cl₂/*n*-pentane); ¹H NMR (300 MHz, CDCl₃): = 4.30 (dd, ³*J* = 11.7 Hz, ³*J* = 3.3 Hz, 1 H, 2-H), 4.56 (d, ³*J* = 11.7 Hz, 1 H, Ar₂C*H*), 6.74–7.16 (m, 33 H, ArH), 9.66 (d, ³*J* = 3.3 Hz, 1 H, 1-H); ¹³C NMR (75.5 MHz, CDCl₃): = 51.3 (d, Ar₂CH), 63.9 (d, C-2), 122.6, 122.9, 123.9, 124.0, 124.1, 127.6, 128.9, 129.0, 129.2, 129.2, 129.4, 129.6 (12 d, Ar), 134.9, 136.0, 136.1, 146.0, 146.6, 147.8 (6 s, Ar), 199.3 (d, C-1).



N-(3,3-Bis(4-(phenyl(2,2,2-trifluoroethyl)amino)phenyl)-1-phenylpropylidene)morpholinium tetrafluoroborate (2l) was obtained from -(N-morpholino)styrene (1l) (51 mg, 0.27 mmol) and (pfa)₂CH⁺ BF₄⁻ (163 mg, 0.272 mmol) as a pale violet solid (179 mg, 83 %) following Procedure A. ¹H NMR (300 MHz, CDCl₃): = 3.55–4.05 (m, 11 H, morpholino-CH₂, 2-H₂, Ar₂CH), 4.17 (q, *J*(H,F) = 8.7 Hz, 4 H, NCH₂CF₃), 6.73–7.50 (m, 23 H, ArH); ¹³C NMR (75.5 MHz, CDCl₃): = 44.1 (t, C-2), 47.0 (d, Ar₂CH), 54.1 (tq, *J*(C,F) = 30.2 Hz, NCH₂CF₃), 53.5, 56.1, 65.9, 66.4 (4 t, morpholino-CH₂), 119.7, 123.1, 124.0 (3 d, Ar), 127.0 (s, Ar), 127.4, 128.5, 129.6, 129.7, 132.7 (5 d, Ar), 131.8, 134.6, 146.8 (3 s, Ar), 188.1 (s, C-1), the signal for CF₃ could not be identified.



3,3-Bis(4-diphenylaminophenyl)-1-phenylpropan-1-one (**3l**) was obtained from -(N-morpholino)styrene (**1l**) (189 mg, 1.00 mmol) and $(dpa)_2CH^+$ BF₄⁻ (589 mg, 1.00 mmol) following Procedure B. Crystallization of the crude product gave a pale green solid (284 mg, 46 %). M.p. 96–97 °C (CH₂Cl₂/*n*-pentane); ¹H NMR (300 MHz, CDCl₃): = 3.67 (d, ³J = 7.2 Hz, 2 H, 2-H), 4.71 (t, ³J = 7.3 Hz, 1 H, Ar₂CH), 6.95–7.93 (m, 33 H, ArH); ¹³C NMR (75.5 MHz, CDCl₃): = 45.07 (t, C-2), 45.09 (d, Ar₂CH), 122.6, 124.0, 124.1, 128.1, 128.6, 129.1, 133.0 (7 d, Ar), 137.2, 138.5, 146.0, 147.8 (4 s, Ar), 198.5 (s, C-1); MS (70 eV, EI):

m/z (%): 620 (22) [M⁺], 501 (100) [Ar₂CH⁺], 251 (10); elemental analysis calcd (%) for C₄₅H₃₆N₂O (620.8): C 87.07, H 5.84, N 4.51; found C 87.26, H 6.00, N 4.42.



N-(2-(Bis(4-dimethylaminophenyl)methyl)cyclopentylidene)methylphenylammonium

tetrafluoroborate (**2o**) was obtained from 1-(methylphenylamino)cyclopentene (**1o**) (72 mg, 0.42 mmol) and (dma)₂CH⁺ BF₄⁻ (142 mg, 0.417 mmol) as a green solid (197 mg, 92 %) following Procedure A. ¹³C NMR (75.5 MHz, CDCl₃): = 20.4 (t), 29.8 (t), 36.0 (t), 40.4 (q, NMe₂), 40.6 (q, NMe₂), 48.8, 51.3, 52.1 (2 d and 1 q, C-2, Ar₂CH, and N⁺CH₃), 112.6, 112.9 (2 d, Ar), 117.4 (s, Ar), 122.9, 123.4, 129.1, 129.4, 130.7 (5 d, Ar), 143.4, 149.6, 150.0 (3 s, Ar), 205.5 (s, C-1); MS (FAB): m/z (%): 426 (2) [M⁺], 253 (100) [Ar₂CH⁺], 237 (11), 174 (13).



2-(Bis(4-diphenylaminophenyl)methyl)cyclohexanone $(3p)^{17}$ was obtained from 1-(methylphenylamino)cyclohexene (1p) (187 mg, 1.00 mmol) and (dpa)₂CH⁺ BF₄⁻ (589 mg, 1.00 mmol) following Procedure B. Crystallization (CH₂Cl₂/*n*-pentane) of the crude product gave **3p** as a pale brown solid (207 mg, 35 %) which showed identical ¹H and ¹³C NMR spectra as the corresponding sample that was obtained from 1-(trimethylsiloxy)cyclohexene and (dpa)₂CH⁺ BF₄⁻ in ref. 17.

1-(2-(Bis(4-dimethylaminophenyl)methyl))cyclohexylidene)-4-methylpiperazinium triflate (2q) was obtained from 1-(1-cyclohexenyl)-4-methylpiperazine (1q) (204 mg, 1.13 mmol) and $(dma)_2CH^+ CF_3SO_3^-$ (456 mg, 1.13 mmol) as a pale green powder (650 mg, 93 %) following Procedure A. ¹H NMR (300 MHz, CDCl₃): = 1.35–1.40 (m, 1 H, 4-H), 1.50–1.56 (m, 1 H, ¹/₂ × CH₂), 1.98–2.10 (m, partially superimposed, 2 H, 3-H and ¹/₂ × CH₂), 2.03 (s, 3 H, NMe), 2.21–2.26 (m, 1 H, 5-H), 2.42–2.58 (m, 2 H, CH₂), 2.76, 2.81 (2 s, 2 × 6 H, NMe₂), 2.90–2.98, 3.12–3.16 (2 m, 2 × 1 H, 6-H₂), 3.73–4.03 (m, 5 H, 2-H and 2 × CH₂), 4.19 (br d, ³*J* = 11.7 Hz, 1 H, Ar₂C*H*), 6.48–6.59, 6.99–7.23 (2 m, 2 × 4 H, ArH); ¹³C NMR (75.5 MHz, CDCl₃): = 18.72 (t, C-4), 28.35 (t, C-5), 30.24 (t, C-6), 30.44 (t, C-3), 40.17/40.25 (2 q, 2 + 1)). NMe₂), 43.80 (q, NMe), 47.09 (d, C-2), 51.51 (d, Ar₂CH), 52.89/53.00 (2 t, CH₂), 54.05/54.22 (2 t, CH₂), 112.73/112.80 (2 d, Ar), 120.63 (q, *J*(C,F) = 321 Hz, CF₃SO₃), 127.03/128.55 (2 s, Ar), 127.99/128.71 (2 d, Ar), 149.39/149.56 (2 s, Ar), 194.86 (s, C-1), signal assignments are based on ¹H, ¹H- and ¹H, ¹³C-COSY experiments.



2-(Bis(4-dimethylaminophenyl)methyl)cyclohexanone¹⁷ (**3q**) was prepared from the iminium salt **2q** after hydrolysis according to Procedure B as pale beige needles (234 mg, 59 %) which showed identical ¹H and ¹³C NMR spectra as the corresponding sample that was obtained from 1-(morpholino)cyclohexene and (dma)₂CH⁺ BF₄⁻ in ref. 17.

2-(Bis(4-(phenyl(2,2,2-trifluoroethyl)amino)phenyl)methyl)pyrrole (5a) was obtained from pyrrole (4a) (179 mg, 2.67 mmol) and (pfa)₂CH⁺ BF₄⁻ (160 mg, 0.267 mmol) following Procedure C. Column chromatography (silica gel/CHCl₃) gave a colorless solid (50 mg, 32 %). ¹H NMR (600 MHz, CDCl₃): = 4.19 (q, ³*J*(H,F) = 8.7 Hz, 4 H, NCH₂CF₃), 5.29 (s, 1 H, Ar₂C*H*), 5.71–5.75 (m, 1 H, 3-H), 6.07 (dd, *J* = 6.0 Hz, *J* = 2.7 Hz, 1 H, 4-H), 6.60–6.64 (m, 1 H, 5-H), 6.80–7.27 (m, 18 H, ArH), 7.73 (br s, 1 H, NH); ¹³C NMR (75.5 MHz, CDCl₃): = 49.1 (d, Ar₂CH), 54.0 (tq, *J*(C,F) = 33 Hz, NCH₂CF₃) 107.8, 108.3, 117.1, 120.8, 121.7, 123.0 (6 d, Ar), 125.2 (sq, *J*(C,F) = 287 Hz, NCH₂CF₃), 129.5, 129.8 (2 d, Ar), 133.8, 137.2,

146.1, 147.3 (4 s, Ar), signal assignments are based on ¹H, ¹H-COSY experiments.



3-(Bis(4-(phenyl(2,2,2-trifluoroethyl)amino)phenyl)methyl)pyrrole (**5c**). A solution of $(pfa)_2CH^+ BF_4^-$ (190 mg, 0.317 mmol) in CH₂Cl₂ (10 mL) was added dropwise to a stirred solution of *N*-(triisopropylsilyl)pyrrole (**4c**) (181 mg, 0.814 mmol) in CH₂Cl₂ (30 mL). After 5 min a solution of tetrabutylammonium fluoride in THF (c = 1 M, 1 mL, 1 mmol) was added. The reaction mixture was subsequently washed with 0.2 M hydrochloric acid (50 mL) and water (50 mL). Separation of the organic layer, drying over CaCl₂, and evaporation of the

solvent in vacuo gave a brownish oil (134 mg, 73 %). Purification by column chromatography (silica gel/CHCl₃) gave **5c** (88 mg, 48 %) and **5a** (17 mg, 9 %). Characterization of **5c**: ¹H NMR (300 MHz, CDCl₃): = 4.27 (q, J(H,F) = 8.8 Hz, 4 H, NCH₂CF₃), 5.31 (s, 1 H, Ar₂CH), 6.05 (br s, 1 H, 5-H), 6.36 (br s, 1 H, 2-H), 6.75 (br s, 1 H, 4-H), 6.91–7.33 (m, 18 H, ArH), 8.02 (br s, 1 H, NH); ¹³C NMR (75.5 MHz, CDCl₃): = 48.6 (d, Ar₂CH), 54.1 (tq, J(C,F) = 32.7 Hz, NCH₂CF₃), 109.2 (d, C-5), 116.9 (d, C-2), 118.1 (d, C-4), 120.5, 121.7, 122.2, 129.4, 129.5 (5 d, Ar), 125.3 (sq, J = 279.4 Hz, NCH₂CF₃), 127.1, 140.2, 145.4, 147.6 (4 s Ar), signal assignments are based on ¹H, ¹H- and ¹H, ¹³C-COSY, gHMBC and NOESY experiments.



3-(Bis(4-(phenyl(2,2,2-trifluoroethyl)amino)phenyl)methyl)indole (**7a**) was obtained from indole (**6a**) (150 mg, 1.28 mmol) and (pfa)₂CH⁺ BF₄⁻ (75 mg, 0.13 mmol) following Procedure C. Column chromatography (silica gel/CHCl₃) gave a colorless solid (55 mg, 70 %). ¹H NMR (300 MHz, CDCl₃): = 4.20 (q, J(H,F) = 8.7 Hz, 4 H, NCH₂CF₃), 5.52 (br s, 1 H, Ar₂CH), 6.55 (br s, 1 H, 2-H), 6.83–7.23 (m, 22 H, ArH), 7.86 (br s, 1 H, NH); ¹³C NMR (75.5 MHz, CDCl₃): = 47.4 (d, Ar₂CH), 54.0 (tq, J(C,F) = 30.2 Hz, NCH₂CF₃), 111.0, 119.4, 119.9 (3d, Ar), 120.1 (s, Ar), 120.9, 121.4, 122.1, 122.4 (4 d, Ar), 123.8 (d, C-2), 125.2 (sq, J(C,F) = 283 Hz, NCH₂CF₃), 126.9 (s, Ar), 129.4, 129.9 (2 d, Ar), 136.7, 138.5, 145.6, 147.5 (4 s, Ar), signal assignments are based on ¹H, ¹H and ¹H, ¹³C-COSY experiments; MS (70 eV, EI): m/z (%): 628 (35) [M]⁺, 513 (100) [Ar₂CH]⁺, 379 (18), 264 (75).



1-Methyl-3-(bis(4-(phenyl(2,2,2-trifluoroethyl)amino)phenyl)methyl)indole (**7b**) was obtained from *N*-methylindole (**6b**) (360 mg, 2.74 mmol) and $(pfa)_2CH^+BF_4^-$ (165 mg, 0.275 mmol) following Procedure C. Column chromatography (silica gel/CHCl₃) gave a colorless solid (38 mg, 22 %). ¹H NMR (300 MHz, CDCl₃): = 3.65 (s, 3 H, NCH₃), 4.21 (q, *J*(H,F) =

8.7 Hz, 4 H, NCH₂CF₃), 5.52 (s, 1 H, Ar₂C*H*), 6.41 (s, 1 H, 2-H), 6.84–7.24 (m, 22 H, ArH), the presence of additional resonances indicate a contamination of the sample with ca. 15 % of **6b**; ¹³C NMR (75.5 MHz, CDCl₃): = 32.7 (q, NCH₃), 47.4 (d, Ar₂CH), 54.1 (tq, *J*(C,F)= 38 Hz, NCH₂CF₃), 109.1, 118.8, 119.9, 120.9, 121.4, 121.6, 122.4 (7 d, Ar), 128.5 (d, C-2), 129.4, 129.9 (2 d, Ar), 138.7, 145.6, 147.5 (3 s, Ar), because of the low signal-to-noise ratio CF₃ and three aromatic carbons could not be identified in the spectra, signal assignments are based on ¹H, ¹³C-COSY experiments; MS (70 eV, EI): m/z (%): 642 (32) [M]⁺, 513 (57) [Ar₂CH]⁺, 393 (44), 264 (42).



1,2-Dimethyl-3-(bis(4-(phenyl(2,2,2-trifluoroethyl)amino)phenyl)methyl)indole (**7c**) was obtained from 1,2-dimethylindole (**6c**) (469 mg, 3.23 mmol) and (pfa)₂CH⁺ BF₄⁻ (194 mg, 0.323 mmol) following Procedure C. Column chromatography (silica gel/CHCl₃) gave a colorless solid (72 mg, 34 %). ¹H NMR (300 MHz, CDCl₃): = 2.19 (s, 3 H, 2-CH₃), 3.57 (s, 3 H, NCH₃), 4.18 (q, *J*(H,F) = 8.7 Hz, 4 H, NCH₂CF₃), 5.61 (s, 1 H, Ar₂CH), 6.83–7.19 (m, 22 H, ArH), the presence of additional resonances indicate a contamination of the sample with ca. 15 % of **6c**; ¹³C NMR (75.5 MHz, CDCl₃): = 10.7 (q, 2-CH₃), 29.5 (q, NCH₃), 46.7 (d, Ar₂CH), 54.0 (tq, *J*(C,F) = 30.2 Hz, NCH₂CF₃), 108.6 (d, Ar), 113.5 (s, Ar), 118.7, 119.5, 120.2, 120.3, 122.0 (5 d, Ar), 125.2 (sq, *J*(C,F) = 287 Hz, NCH₂CF₃), 127.3 (s, Ar), 129.3, 130.2 (2 d, Ar), 133.7, 136.7, 139.1, 145.3, 147.7 (5 s, Ar); MS (70 eV, EI): *m/z* (%): 657 (38) [M⁺], 513 (100) [Ar₂CH⁺], 407 (61), 264 (14).



B. Concentrations and rate constants of the individual kinetic runs (Tables S1-S55)

Table S1. 1-(*N*-Pyrrolidino)cyclopentene (**1a**) and (lil)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 640$ nm (Stopped flow).

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	No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	$T/^{\circ}C$	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
	011200-F	8.456×10^{-6}	1.436×10^{-4}	17	20.0	1.251×10^5
	011200-В	8.456×10^{-6}	2.871×10^{-4}	34	20.0	1.245×10^5
	011200-С	8.456×10^{-6}	4.307×10^{-4}	51	20.0	1.307×10^5
	011200-D	8.456×10^{-6}	5.743×10^{-4}	68	20.0	1.203×10^5
	011200-Е	8.456×10^{-6}	7.178×10^{-4}	85	20.0	1.290×10^5

 $<k_2>(20 \ ^{\circ}C) = (1.259 \pm 0.037) \times 10^5 \ M^{-1} \ s^{-1}$

Table S2. 1-(*N*-Pyrrolidino)cyclopentene (**1a**) and (jul)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 640$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
011200-М	7.130×10^{-6}	1.425×10^{-4}	20	20.0	3.387×10^{5}
011200-О	7.130×10^{-6}	2.851×10^{-4}	40	20.0	3.171×10^5
011200-I	7.130×10^{-6}	4.276×10^{-4}	60	20.0	3.182×10^5
011200-J	7.130×10^{-6}	5.702×10^{-4}	80	20.0	3.459×10^5
011200-К	7.130×10^{-6}	7.127×10^{-4}	100	20.0	3.419×10^5

 $<k_2>(20 \text{ °C}) = (3.324 \pm 0.122) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

Table S3. 1-(*N*-Pyrrolidino)cyclopentene (**1a**) and $(thq)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 628$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
100501-D	5.099×10^{-6}	1.482×10^{-5}	3.4	20.0	4.707×10^{6}
100501-B	5.099×10^{-6}	2.351×10^{-5}	5.1	20.0	4.112×10^{6}
100501-E	5.099×10^{-6}	3.220×10^{-5}	6.8	20.0	4.698×10^6
100501-A	5.099×10^{-6}	4.088×10^{-5}	8.5	20.0	4.176×10^6
100501-C	5.099×10^{-6}	4.957×10^{-5}	10	20.0	4.693×10^6

 $< k_2 > (20 \ ^{\circ}\text{C}) = (4.477 \pm 0.273) \times 10^6 \ \text{M}^{-1} \ \text{s}^{-1}$

Table S4. 1-(*N*-Pyrrolidino)cyclohexene (**1b**) and $(lil)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 640$ nm (Schölly).

$/ M^{-1} s^{-1}$
581×10^2
786×10^2
234×10^2
304×10^3
006×10^{3}
786 > 234 > 304 > 006 >



$$k_2(20 \text{ °C}) = (1.482 \pm 0.167) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$$

Table S5. 1-(*N*-Pyrrolidino)cyclohexene (**1b**) and $(jul)_2CH^+BF_4^-$ in CH_2Cl_2 at = 640 nm (Stopped flow).

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	No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	$T / ^{\circ}\mathrm{C}$	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
	021200-F	6.698×10^{-6}	1.386×10^{-4}	21	20.0	4.183×10^{4}
	021200-I	6.698×10^{-6}	2.079×10^{-4}	31	20.0	4.722×10^4
	021200-В	6.698×10^{-6}	2.772×10^{-4}	41	20.0	4.451×10^4
	021200-J	6.698×10^{-6}	3.464×10^{-4}	52	20.0	4.841×10^4
	021200-С	6.698×10^{-6}	4.157×10^{-4}	62	20.0	4.732×10^4

 $<k_2>(20 \ ^{\circ}C) = (4.586 \pm 0.239) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

Table S6. 1-(*N*-Pyrrolidino)cyclohexene (**1b**) and $(thq)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 628$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
030501-C	6.119×10^{-6}	6.565×10^{-5}	11	20.0	$7.508 imes 10^5$
030501-B	6.119×10^{-6}	1.313×10^{-4}	22	20.0	7.000×10^5
030501-A	6.119×10^{-6}	2.626×10^{-4}	43	20.0	7.377×10^{5}

 $< k_2 > (20 \ ^{\circ}\text{C}) = (7.295 \pm 0.215) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

Table S7. 1-(*N*-Pyrrolidino)cyclohexene (**1b**) and $(dma)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 613$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
160501-D	5.033×10^{-6}	1.244×10^{-5}	3.0	20.0	4.972×10^{6}
160501-B	5.033×10^{-6}	2.241×10^{-5}	5.0	20.0	5.438×10^6
160501-C	5.033×10^{-6}	3.238×10^{-5}	6.9	20.0	$5.755 imes 10^6$
160501-A	5.033×10^{-6}	4.733×10^{-5}	9.9	20.0	5.172×10^{6}

 $< k_2 > (20 \ ^{\circ}\text{C}) = (5.334 \pm 0.294) \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$

Table S8. 1-(*N*-Piperidino)cyclopentene (1c) and (lil)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 640$ nm (Schölly).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	Conv. / %	T/°C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
100500.PA1	4.658×10^{-5}	6.218×10^{-4}	13	74	-77.7	2.301×10^2
100500.PA2	5.053×10^{-5}	6.745×10^{-4}	13	60	-71.8	3.297×10^2
100500.PA3	4.417×10^{-5}	5.527×10^{-4}	13	46	-61.7	5.338×10^2
100500.PA5	4.738×10^{-5}	4.744×10^{-4}	10	74	-51.9	8.379×10^2
100500.PA7	5.150×10^{-5}	2.644×10^{-4}	5	61	-41.5	1.383×10^3
100500.PA8	4.031×10^{-5}	2.690×10^{-4}	7	35	-31.7	2.196×10^3



 $k_2(20 \ ^\circ\text{C}) = (1.146 \pm 0.067) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

Table S9. 1-(*N*-Piperidino)cyclopentene (1c) and (jul)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 640$ nm (Schölly).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	Conv. / %	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
161100.PA0	2.200×10^{-5}	1.172×10^{-3}	53	65	-80.6	6.366×10^{2}
161100.PA1	1.975×10^{-5}	4.209×10^{-4}	21	60	-78.8	7.672×10^2
161100.PA2	2.636×10^{-5}	5.617×10^{-4}	21	41	-75.3	8.529×10^2
161100.PA3	2.260×10^{-5}	3.854×10^{-4}	17	33	-70.0	1.141×10^3
161100.PA4	4.256×10^{-5}	4.883×10^{-4}	11	73	-68.1	1.296×10^3
161100.PA5	2.835×10^{-5}	3.253×10^{-4}	11	55	-60.5	2.145×10^3
161100.PA6	3.815×10^{-5}	3.127×10^{-4}	8	48	-53.1	2.964×10^3
161100.PA7	3.427×10^{-5}	1.685×10^{-4}	5	53	-48.7	3.665×10^3



Eyring parameters:

Arrhenius parameters:

$$\Delta H^{\ddagger} = 17.969 \pm 0.556 \text{ kJ mol}^{-1} \qquad E_{a} = 19.695 \pm 0.561 \text{ kJ mol}^{-1}$$
$$\Delta S^{\ddagger} = -94.280 \pm 2.707 \text{ J mol}^{-1} \text{ K}^{-1} \qquad \ln A = 18.757 \pm 0.329$$
$$r^{2} = 0.9943 \qquad r^{2} = 0.9952$$

$$k_2(20 \text{ °C}) = (4.565 \pm 0.444) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$$

Table S10. 1-(*N*-Piperidino)cyclopentene (1c) and $(thq)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 624$ nm (Stopped flow).

No.	$[El]_0 / M$	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
171100-A	1.963×10^{-5}	2.034×10^{-4}	10	20.0	4.333×10^{5}
171100-В	1.963×10^{-5}	4.067×10^{-4}	21	20.0	4.429×10^5
171100-С	1.963×10^{-5}	6.101×10^{-4}	31	20.0	4.455×10^5
171100-D	1.963×10^{-5}	8.135×10^{-4}	41	20.0	4.451×10^5
171100-Е	1.963×10^{-5}	1.017×10^{-4}	52	20.0	4.538×10^5

 $<k_2>(20 \ ^{\circ}\text{C}) = (4.441 \pm 0.066) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

Table S11. 1-(*N*-Piperidino)cyclopentene (**1c**) and $(dma)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 613$ nm (Stopped flow).

No.	$[El]_0 / M$	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
220301-C	3.351×10^{-6}	1.845×10^{-5}	6	20.0	3.766×10^{6}
220301-В	3.351×10^{-6}	3.074×10^{-5}	9	20.0	3.605×10^6
220301-D	3.351×10^{-6}	4.304×10^{-5}	13	20.0	3.703×10^{6}
220301-A	3.351×10^{-6}	6.149×10^{-5}	18	20.0	3.656×10^6

 $<k_2>(20 \text{ °C}) = (3.683 \pm 0.059) \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$

No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	Conv. / %	T/°C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
040500.PA0	3.707×10^{-5}	1.173×10^{-3}	32	84	-73.7	3.151
040500.PA1	3.123×10^{-5}	7.904×10^{-4}	25	84	-62.6	6.784
040500.PA2	2.802×10^{-5}	4.433×10^{-4}	16	86	-52.1	1.385×10^1
040500.PA3	2.455×10^{-5}	7.768×10^{-4}	32	93	-40.8	2.848×10^1
040500.PA5	2.623×10^{-5}	6.639×10^{-4}	25	84	-30.8	4.988×10^1
040500.PA6	2.382×10^{-5}	4.522×10^{-4}	19	90	-20.4	8.150×10^1
040500.PA7	2.468×10^{-5}	3.904×10^{-4}	16	66	-10.5	1.358×10^2
040500.PA8	2.743×10^{-5}	6.941×10^{-4}	25	71	0.2	2.135×10^2

Table S12. 1-(*N*-Morpholino)cyclopentene (**1e**) and (lil)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 640$ nm (Schölly).



 $k_2(20 \text{ °C}) = (4.703 \pm 0.075) \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$

Table S13. 1-(*N*-Morpholino)cyclopentene (1e) and $(jul)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 640$ nm (Schölly).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	Conv. / %	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
KK050520.PA2	2.594×10^{-5}	3.219×10^{-4}	12	34	-67.6	2.523×10^1
KK050521.PA2	2.963×10^{-5}	3.677×10^{-4}	12	83	-58.2	4.650×10^1
KK050522.PA2	2.429×10^{-5}	3.014×10^{-4}	12	72	-58.1	4.741×10^{1}
KK050523.PA2	1.614×10^{-5}	1.001×10^{-4}	6	79	-48.0	8.000×10^1
KK050524.PA2	2.765×10^{-5}	3.431×10^{-4}	12	73	-38.2	1.528×10^2
KK050525.PA2	2.551×10^{-5}	2.216×10^{-4}	9	75	-28.3	2.407×10^2
KK090523.PA0	1.423×10^{-5}	1.360×10^{-4}	10	64	-23.3	3.084×10^2



$$r^2 = 0.9986$$

 $k_2(20 \text{ °C}) = (1.760 \pm 0.081) \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$

No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	Conv. / %	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
080520.PA1	2.363×10^{-5}	2.122×10^{-4}	9	85	-71.7	4.398×10^2
080520.PA2	3.623×10^{-5}	4.163×10^{-4}	12	66	-61.2	8.128×10^2
080520.PA4	2.476×10^{-5}	1.779×10^{-4}	7	66	-50.5	1.345×10^3
080520.PA5	4.099×10^{-5}	2.265×10^{-4}	6	56	-40.6	1.846×10^3
080520.PA6	2.423×10^{-5}	1.045×10^{-4}	4	51	-30.9	3.215×10^3
080520.PA7	3.374×10^{-5}	1.118×10^{-4}	3	47	-20.7	4.738×10^3

Table S14. 1-(*N*-Morpholino)cyclopentene (**1e**) and $(thq)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 640$ nm (Schölly).



$$k_2(20 \text{ °C}) = (1.705 \pm 0.136) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$$

Table S15. 1-(*N*-Morpholino)cyclopentene (1e) and $(dma)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 613$ nm (Stopped flow).

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	No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T/°C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
	190401-B	5.080×10^{-6}	8.628×10^{-5}	17	20.0	2.323×10^5
	190401-A	5.080×10^{-6}	1.726×10^{-4}	34	20.0	2.557×10^5
	190401-C	5.080×10^{-6}	2.588×10^{-4}	51	20.0	2.421×10^5
	190401-D	5.080×10^{-6}	3.451×10^{-4}	68	20.0	2.446×10^5

 $< k_2 > (20 \text{ °C}) = (2.437 \pm 0.083) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	$T / ^{\circ}\mathrm{C}$	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
150501-C	4.911×10^{-6}	3.733×10^{-5}	8	20.0	9.585×10^{5}
150501-B	4.911×10^{-6}	7.466×10^{-5}	15	20.0	9.952×10^5
150501-Е	4.911×10^{-6}	1.120×10^{-4}	23	20.0	9.585×10^5
150501-F	4.911×10^{-6}	1.493×10^{-4}	30	20.0	9.669×10^5
150501-D	4.911×10^{-6}	1.867×10^{-4}	38	20.0	9.742×10^5

Table S16. 1-(*N*-Morpholino)cyclopentene (1e) and (mpa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 622$ nm (Stopped flow).

 $< k_2 > (20 \ ^{\circ}C) = (9.707 \pm 0.136) \times 10^5 \ M^{-1} \ s^{-1}$

Table S17. 1,2,5,6-Tetrahydro-1-methyl-4-(*N*-morpholino)pyridine (**1g**) and (dma)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 600$ nm (Schölly).

No.	[<i>El</i>] ₀ / M	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	Conv. / %	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
AO-162.2	2.63×10^{-5}	$1.78 imes 10^{-4}$	6.8	64	-69.8	7.074×10^1
AO-162.3	3.02×10^{-5}	$2.03 imes 10^{-4}$	6.7	72	-61.9	1.166×10^2
AO-162.4	2.84×10^{-5}	1.92×10^{-4}	6.8	72	-48.7	3.886×10^2
AO-162.5	2.68×10^{-5}	1.81×10^{-4}	6.8	81	-38.3	5.568×10^2
AO-162.6	$2.60 imes 10^{-5}$	1.75×10^{-4}	6.7	87	-28.2	1.068×10^3



 $k_2(20 \text{ °C}) = (1.010 \pm 0.222) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

No.	$[El]_0 / \mathbf{M}$	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
270801-Е	4.962×10^{-6}	5.488×10^{-5}	11	20.0	7.722×10^5
270801-D	4.962×10^{-6}	1.098×10^{-4}	22	20.0	$7.909 imes 10^5$
270801-A	4.962×10^{-6}	1.646×10^{-4}	33	20.0	7.380×10^5
270801-В	4.962×10^{-6}	2.195×10^{-4}	44	20.0	7.250×10^5
270801-C	4.962×10^{-6}	$2.744{\times}~10^{-4}$	55	20.0	6.960×10^5

Table S18. (*E*)-1-(*N*-Morpholino)propene (**1h**) and $(dpa)_2CH^+ BF_4^-$ in CH_2Cl_2 at $\lambda = 672$ nm (Stopped flow).

 $< k_2 > (20 \ ^{\circ}\text{C}) = (7.444 \pm 0.338) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

Table S19. (*Z*)-1-(*N*-Morpholino)propene^[a] (1i) and $(dpa)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 672$ nm (Stopped flow).

No.	$[El]_0 / \mathbf{M}$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	T∕°C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
170801-B	4.650×10^{-6}	5.394×10^{-5}	12	20.0	1.148×10^{6}
170801-C	4.650×10^{-6}	1.079×10^{-4}	23	20.0	1.036×10^6
170801-A	4.650×10^{-6}	1.618×10^{-4}	35	20.0	9.775×10^5
170801-D	4.650×10^{-6}	2.157×10^{-4}	46	20.0	1.141×10^{6}
170801-E	4.650×10^{-6}	2.697×10^{-4}	58	20.0	1.046×10^6

[a] contaminated by 11 % of (*E*)-1-(*N*-morpholino)propene (**1h**) (¹H NMR)

 $<k_2>(20 \text{ °C}) = (1.070 \pm 0.065) \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$

Table S20. (*N*-Morpholino)isobutylene (**1j**) and $(dma)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 600$ nm (Schölly).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	Conv. / %	T/°C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
050701.PA1	6.349×10^{-5}	2.314×10^{-3}	37	81	-73.6	2.133
050701.PA2	6.314×10^{-5}	2.301×10^{-3}	37	77	-59.6	5.751
050701.PA3	5.656×10^{-5}	3.298×10^{-3}	58	57	-49.8	1.039×10^1
050701.PA4	5.801×10^{-5}	2.537×10^{-3}	44	61	-40.7	1.801×10^1
050701.PA5	6.162×10^{-5}	1.797×10^{-3}	29	51	-29.7	3.048×10^1



 $k_2(20 \text{ °C}) = (2.528 \pm 0.095) \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$

Table S21. (*N*-Morpholino)isobutylene (**1j**) and (mpa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 622$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	T / °C	$k_{\rm eff}$ / s ⁻¹
090701-Е	$9.735 imes 10^{-6}$	3.255×10^{-4}	33	20.0	1.250
090701-D	9.735×10^{-6}	6.509×10^{-4}	67	20.0	2.223
090701-A	9.735×10^{-6}	9.764×10^{-4}	100	20.0	3.274
090701-B	9.735×10^{-6}	1.302×10^{-3}	134	20.0	4.322
090701-C	9.735×10^{-6}	1.627×10^{-3}	167	20.0	5.611



 $< k_2 > (20 \ ^{\circ}\text{C}) = 3.325 \times 10^3 \text{ L mol}^{-1} \text{ s}^{-1}$ $< k_{-2} > (20 \ ^{\circ}\text{C}) = 8.931 \times 10^{-2} \text{ s}^{-1}$ $K(20 \ ^{\circ}\text{C}) = 3.723 \times 10^4 \text{ L mol}^{-1}$

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
030701-Е	1.041×10^{-5}	7.322×10^{-5}	7	20.0	2.312×10^4
030701-D	1.041×10^{-5}	1.464×10^{-4}	14	20.0	2.368×10^4
030701-A	1.041×10^{-5}	2.197×10^{-4}	21	20.0	2.383×10^4
030701-B	1.041×10^{-5}	2.929×10^{-4}	28	20.0	2.434×10^4
030701-C	1.041×10^{-5}	3.661×10^{-4}	35	20.0	2.557×10^4

Table S22. (*N*-Morpholino)isobutylene (**1j**) and $(dpa)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 672$ nm (Stopped flow).

 $< k_2 > (20 \ ^{\circ}\text{C}) = (2.411 \pm 0.083) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

Table S23. (*N*-Morpholino)isobutylene (**1j**) and $(mfa)_2CH^+ BF_4^-$ in CH_2Cl_2 at $\lambda = 593$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
 030701-J	1.033×10^{-5}	7.365×10^{-5}	7	20.0	1.251×10^{5}
030701-I	1.033×10^{-5}	1.473×10^{-4}	14	20.0	1.273×10^5
030701-F	1.033×10^{-5}	2.209×10^{-4}	21	20.0	1.283×10^5
030701-G	1.033×10^{-5}	2.946×10^{-4}	29	20.0	1.298×10^5
030701-Н	1.033×10^{-5}	3.682×10^{-4}	36	20.0	1.311×10^5

 $<k_2>(20 \text{ °C}) = (1.283 \pm 0.021) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

Table S24. (*N*-Morpholino)isobutylene (**1j**) and (pfa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 601$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
030701-O	$6.030 imes 10^{-6}$	6.246×10^{-5}	10	20.0	4.087×10^{5}
030701-N	6.030×10^{-6}	1.249×10^{-4}	21	20.0	4.260×10^5
030701-K	6.030×10^{-6}	1.874×10^{-4}	31	20.0	4.195×10^5
030701-L	6.030×10^{-6}	2.498×10^{-4}	41	20.0	4.420×10^5

 $< k_2 > (20 \ ^{\circ}\text{C}) = (4.240 \pm 0.121) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

 $k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$ No. $T / ^{\circ}C$ $[El]_0 / M$ $[Nuc]_0 / M$ $[Nuc]_0/[El]_0$ 1.626×10^{4} 6.099×10^{-6} 5.041×10^{-4} 120701-I 83 20.0 7.561×10^{-4} $1.476 imes 10^4$ 6.099×10^{-6} 120701-K 124 20.0 6.099×10^{-6} 1.008×10^{-3} 1.371×10^{4} 120701-J 165 20.0

Table S25. (*E*)- -(*N*-Morpholino)styrene (**1k**) and (mpa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 622$ nm (Stopped flow).

 $< k_2 > (20 \ ^{\circ}\text{C}) = (1.491 \pm 0.105) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

Table S26. (*E*)- -(*N*-Morpholino)styrene (**1k**) and $(dpa)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 672$ nm (Stopped flow).

No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	T∕°C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
120701-Е	$6.580 imes 10^{-6}$	1.008×10^{-4}	15	20.0	2.118×10^{5}
120701-D	6.580×10^{-6}	2.016×10^{-4}	31	20.0	2.247×10^5
120701-A	6.580×10^{-6}	3.024×10^{-4}	46	20.0	2.218×10^5
120701-В	6.580×10^{-6}	4.033×10^{-4}	61	20.0	2.258×10^5
120701-C	6.580×10^{-6}	5.041×10^{-4}	77	20.0	2.212×10^5

 $< k_2 > (20 \ ^{\circ}C) = (2.210 \pm 0.050) \times 10^5 \ M^{-1} \ s^{-1}$

Table S27. (*E*)- -(*N*-Morpholino)styrene (**1k**) and (mfa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 593$ nm (Stopped flow).

No.	$[El]_0$ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
120701-G	9.005×10^{-6}	5.041×10^{-5}	6	20.0	1.128×10^{6}
120701-F	9.005×10^{-6}	1.008×10^{-4}	11	20.0	1.075×10^6
120701-Н	9.005×10^{-6}	2.016×10^{-4}	22	20.0	9.943×10^5

 $< k_2 > (20 \ ^{\circ}\text{C}) = (1.066 \pm 0.055) \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$

 $k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$ No. $T / ^{\circ}C$ $[El]_0 / M$ $[Nuc]_0 / M$ $[Nuc]_0/[El]_0$ 3.995×10^{6} 6.823×10^{-6} 5.041×10^{-5} 120701-M 7 20.0 6.823×10^{-6} 7.057×10^{-5} 3.675×10^6 120701-N 10 20.0 6.823×10^{-6} 1.008×10^{-4} 3.481×10^{6} 120701-L 15 20.0

Table S28. (*E*)- -(*N*-Morpholino)styrene (**1k**) and (pfa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 601$ nm (Stopped flow).

 $< k_2 > (20 \ ^{\circ}\text{C}) = (3.717 \pm 0.212) \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$

Table S29. -(*N*-Morpholino)styrene (11) and $(thq)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 640$ nm (Schölly).

No.	[<i>El</i>] ₀ / M	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	Conv. / %	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
010601.PA2	2.987×10^{-5}	4.332×10^{-4}	15	78	20.0	2.347×10^{1}
010601.PA0	3.442×10^{-5}	9.982×10^{-4}	29	79	20.0	2.330×10^1
010601.PA1	2.929×10^{-5}	2.124×10^{-3}	73	70	20.0	2.356×10^1

 $< k_2 > (20 \ ^{\circ}\text{C}) = (2.344 \pm 0.011) \times 10^1 \text{ M}^{-1} \text{ s}^{-1}$

Table S30. -(*N*-Morpholino)styrene (11) and $(dma)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 600$ nm (Schölly).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	Conv. / %	T/°C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
310501.PA1	3.767×10^{-5}	1.023×10^{-3}	27	68	-50.0	1.692
310501.PA2	4.421×10^{-5}	1.200×10^{-3}	27	79	-41.2	3.809
310501.PA3	3.919×10^{-5}	8.513×10^{-4}	22	80	-29.9	8.979
310501.PA4	3.160×10^{-5}	5.148×10^{-4}	16	76	-20.6	1.818×10^1
310501.PA5	3.088×10^{-5}	3.354×10^{-4}	11	81	-10.3	3.519×10^{1}



Eyring parameters:	Arrhenius parameters:
$\Delta H^{\ddagger} = 35.203 \pm 0.413 \text{ kJ mol}^{-1}$	$E_{\rm a} = 37.213 \pm 0.396 \text{ kJ mol}^{-1}$
$\Delta S^{\ddagger} = -80.197 \pm 1.709 \text{ J mol}^{-1} \text{ K}^{-1}$	$\ln A = 20.604 \pm 0.197$
$r^2 = 0.9996$	$r^2 = 0.9997$

$$k_2(20 \text{ °C}) = (2.111 \pm 0.076) \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$$

Table S31. -(*N*-Morpholino)styrene (11) and $(mpa)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 622$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
300501-Е	1.347×10^{-5}	$2.220 imes 10^{-4}$	17	20.0	1.540×10^{3}
300501-D	1.347×10^{-5}	4.440×10^{-4}	33	20.0	1.552×10^3
300501-F	1.347×10^{-5}	6.661×10^{-4}	49	20.0	1.793×10^3
300501-C	1.347×10^{-5}	8.881×10^{-4}	66	20.0	1.731×10^3
300501-В	1.347×10^{-5}	1.110×10^{-3}	82	20.0	1.776×10^3

 $< k_2 > (20 \text{ °C}) = (1.678 \pm 0.110) \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$

Table S32. -(*N*-Morpholino)styrene (11) and $(dpa)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 672$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
290501-Е	1.380×10^{-5}	1.073×10^{-4}	8	20.0	1.327×10^4
290501-В	1.380×10^{-5}	2.145×10^{-4}	16	20.0	1.345×10^4
290501-С	1.380×10^{-5}	3.218×10^{-4}	23	20.0	1.353×10^4
290501-D	1.380×10^{-5}	4.290×10^{-4}	31	20.0	$1.377 imes 10^4$
290501-A	1.380×10^{-5}	5.363×10^{-4}	39	20.0	$1.375 imes 10^4$

 $<k_2>(20 \text{ °C}) = (1.355 \pm 0.019) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
290501-J	7.292×10^{-6}	1.073×10^{-4}	15	20.0	6.428×10^{4}
290501-F	7.292×10^{-6}	$2.145 imes 10^{-4}$	29	20.0	$6.512 imes 10^4$
290501-I	7.292×10^{-6}	3.218×10^{-4}	44	20.0	$6.737 imes 10^4$
290501-Н	7.292×10^{-6}	4.290×10^{-4}	59	20.0	$6.763 imes 10^4$
290501-G	7.292×10^{-6}	5.363×10^{-4}	74	20.0	$6.940 imes 10^4$

Table S33. -(*N*-Morpholino)styrene (11) and $(mfa)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 593$ nm (Stopped flow).

 $< k_2 > (20 \ ^{\circ}\text{C}) = (6.676 \pm 0.184) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

Table S34. -(*N*-Morpholino)styrene (11) and $(pfa)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 601$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
290501-O	5.251×10^{-6}	1.073×10^{-4}	20	20.0	2.403×10^{5}
290501-К	5.251×10^{-6}	2.145×10^{-4}	41	20.0	2.352×10^5
290501-N	5.251×10^{-6}	3.218×10^{-4}	61	20.0	2.515×10^5
290501-M	5.251×10^{-6}	4.290×10^{-4}	82	20.0	2.573×10^5
290501-L	5.251×10^{-6}	5.363×10^{-4}	102	20.0	2.611×10^5

 $<k_2>(20 \text{ °C}) = (2.491 \pm 0.099) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

Table S35. Ethyl (*E*)-3-(*N*-morpholino)acrylate (**1m**) and (pfa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 601$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	<i>T</i> / °C	$k_{\rm eff}$ / s ⁻¹
240701-Е	6.397×10^{-6}	2.234×10^{-4}	35	20.0	6.459
240701-D	6.397×10^{-6}	4.468×10^{-4}	70	20.0	1.109×10^1
240701-A	6.397×10^{-6}	6.702×10^{-4}	105	20.0	1.490×10^1
240701-В	6.397×10^{-6}	8.936×10^{-4}	140	20.0	2.008×10^1
240701-C	6.397×10^{-6}	1.117×10^{-3}	175	20.0	2.452×10^1



Table S36. Ethyl (*E*)-3-(dimethylamino)acrylate (**1n**) and (pfa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 601$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
240801-В	5.397×10^{-6}	1.589×10^{-4}	30	20.0	1.128×10^5
240801-С	5.397×10^{-6}	2.384×10^{-4}	44	20.0	1.027×10^5
240801-D	5.397×10^{-6}	3.179×10^{-4}	59	20.0	1.039×10^5
240801-Е	5.397×10^{-6}	3.974×10^{-4}	74	20.0	1.064×10^5
240801-G	5.397×10^{-6}	5.563×10^{-4}	103	20.0	1.112×10^{5}

 $< k_2 > (20 \ ^{\circ}\text{C}) = (1.074 \pm 0.040) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

Table S37. 1-(Methylphenylamino)cyclopentene (**10**) and (lil)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 640$ nm (Schölly).

No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	Conv. / %	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
210601.PA0	3.690×10^{-5}	1.106×10^{-3}	30	76	-47.8	2.182
210601.PA1	3.383×10^{-5}	1.268×10^{-3}	38	74	-40.5	4.464
210601.PA2	2.473×10^{-5}	8.473×10^{-4}	34	70	-30.9	8.740
210601.PA3	3.068×10^{-5}	6.898×10^{-4}	23	65	-20.4	1.820×10^1
210601.PA4	2.584×10^{-5}	6.640×10^{-4}	26	59	-11.2	3.392×10^1



 $k_2(20 \text{ °C}) = (2.017 \pm 0.137) \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$

Table S38. 1-(Methylphenylamino)cyclopentene (10) and $(jul)_2CH^+ BF_4^-$ in CH_2Cl_2 at $\lambda = 642$ nm (Stopped flow).

No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
180601-L	5.131×10^{-6}	2.036×10^{-4}	40	20.0	4.175×10^{2}
180601-K	5.131×10^{-6}	3.394×10^{-4}	66	20.0	4.330×10^2
180601-M	5.131×10^{-6}	6.787×10^{-4}	132	20.0	4.435×10^2

 $< k_2 > (20 \ ^{\circ}\text{C}) = (4.313 \pm 0.107) \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$

Table S39. 1-(Methylphenylamino)cyclopentene (**10**) and $(thq)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 628$ nm (Stopped flow).

_						
	No.	$[El]_0 / \mathbf{M}$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
	180601-J	6.037×10^{-6}	1.357×10^{-4}	23	20.0	4.835×10^{3}
	180601-G	6.037×10^{-6}	2.036×10^{-4}	34	20.0	5.101×10^3
	180601-H	6.037×10^{-6}	2.715×10^{-4}	45	20.0	5.065×10^3
	180601-I	6.037×10^{-6}	3.394×10^{-4}	56	20.0	5.044×10^3

 $< k_2 > (20 \ ^{\circ}\text{C}) = (5.011 \pm 0.104) \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$

No.	$[El]_0 / \mathbf{M}$	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
180601-F	7.149×10^{-6}	6.014×10^{-5}	8	20.0	4.664×10^{4}
180601-D	7.149×10^{-6}	1.203×10^{-4}	17	20.0	$4.808 imes 10^4$
180601-A	7.149×10^{-6}	1.804×10^{-4}	25	20.0	5.082×10^4
180601-B	7.149×10^{-6}	2.406×10^{-4}	34	20.0	5.152×10^4
180601-C	7.149×10^{-6}	3.007×10^{-4}	42	20.0	5.097×10^4

Table S40. 1-(Methylphenylamino)cyclopentene (**1o**) and $(dma)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 613$ nm (Stopped flow).

 $< k_2 > (20 \ ^{\circ}C) = (4.961 \pm 0.190) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

Table S41. 1-(Methylphenylamino)cyclopentene (**1o**) and (mpa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 622$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
130601-K	4.997×10^{-6}	6.464×10^{-5}	13	20.0	3.334×10^5
130601-G	4.997×10^{-6}	1.293×10^{-4}	26	20.0	3.265×10^5
130601-Н	4.997×10^{-6}	1.939×10^{-4}	39	20.0	3.179×10^5
130601-I	4.997×10^{-6}	2.586×10^{-4}	52	20.0	3.141×10^5
130601-J	4.997×10^{-6}	3.232×10^{-4}	65	20.0	3.125×10^{5}

 $<k_2>(20 \text{ °C}) = (3.209 \pm 0.079) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

Table S42. 1-(Methylphenylamino)cyclopentene (**10**) and $(dpa)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 672$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
130601-C	$5.791 imes 10^{-6}$	3.232×10^{-5}	6	20.0	2.815×10^6
130601-F	5.791×10^{-6}	5.818×10^{-5}	10	20.0	2.738×10^6
130601-B	5.791×10^{-6}	6.464×10^{-5}	11	20.0	2.853×10^6
130601-Е	5.791×10^{-6}	1.293×10^{-4}	22	20.0	2.646×10^6

 $< k_2 > (20 \ ^{\circ}\text{C}) = (2.763 \pm 0.079) \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$

No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	Conv. / %	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
070501-02	1.351×10^{-5}	3.104×10^{-4}	23	86	20.0	1.136×10^{2}
070501-03	1.907×10^{-5}	8.214×10^{-4}	43	95	20.0	1.031×10^2
070501-01	1.736×10^{-5}	7.974×10^{-4}	46	95	20.0	1.034×10^2
070501-04	2.750×10^{-5}	1.579×10^{-3}	57	93	20.0	9.521×10^1

Table S43. 1-(Methylphenylamino)cyclohexene (1p) and $(thq)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 628$ nm (J&M).

 $< k_2 > (20 \ ^{\circ}\text{C}) = (1.038 \pm 0.065) \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$

Table S44. 1-(Methylphenylamino)cyclohexene (**1p**) and (dma)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 613$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_{\rm eff}$ / s ⁻¹
020501-Е	8.890×10^{-6}	2.141×10^{-4}	24	20.0	$2.023 imes 10^{-1}$
020501-B	8.890×10^{-6}	4.282×10^{-4}	48	20.0	4.127×10^{-1}
020501-C	8.890×10^{-6}	6.424×10^{-4}	72	20.0	6.359×10^{-1}
020501-D	8.890×10^{-6}	8.565×10^{-4}	96	20.0	8.813×10^{-1}
020501-A	8.890×10^{-6}	1.071×10^{-3}	120	20.0	1.090



 $< k_2 > (20 \ ^{\circ}\text{C}) = 1.048 \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$ $< k_{-2} > (20 \ ^{\circ}\text{C}) = 2.860 \times 10^{-2} \text{ s}^{-1}$ $K(20 \ ^{\circ}\text{C}) = 3.664 \times 10^4 \text{ M}^{-1}$

No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
300401-L	4.256×10^{-6}	2.976×10^{-4}	70	20.0	8.368×10^{3}
300401-O	4.256×10^{-6}	3.968×10^{-4}	93	20.0	8.503×10^3
300401-K	4.256×10^{-6}	4.960×10^{-4}	117	20.0	8.367×10^3
300401-N	4.256×10^{-6}	6.945×10^{-4}	163	20.0	8.407×10^3
300401-M	4.256×10^{-6}	9.921×10^{-4}	233	20.0	8.492×10^3

Table S45. 1-(Methylphenylamino)cyclohexene (**1p**) and (mpa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 622$ nm (Stopped flow).

 $< k_2 > (20 \ ^{\circ}\text{C}) = (8.427 \pm 0.059) \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$

Table S46. 1-(Methylphenylamino)cyclohexene (**1p**) and (dpa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 672$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
300401-J	$5.975 imes 10^{-6}$	9.921×10^{-5}	17	20.0	8.602×10^{4}
300401-I	5.975×10^{-6}	1.984×10^{-4}	33	20.0	8.720×10^4
300401-Н	5.975×10^{-6}	2.976×10^{-4}	50	20.0	8.705×10^4
300401-G	5.975×10^{-6}	3.968×10^{-4}	66	20.0	8.633×10^4
300401-F	5.975×10^{-6}	4.960×10^{-4}	83	20.0	8.519×10^4

 $<k_2>(20 \text{ °C}) = (8.636 \pm 0.073) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

Table S47. 1-(Methylphenylamino)cyclohexene (**1p**) and (mfa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 593$ nm (Stopped flow).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
300401-B	5.410×10^{-6}	9.921×10^{-5}	18	20.0	3.382×10^{5}
300401-A	5.410×10^{-6}	1.984×10^{-4}	37	20.0	3.357×10^5
300401-C	5.410×10^{-6}	2.976×10^{-4}	55	20.0	3.376×10^5
300401-D	5.410×10^{-6}	3.968×10^{-4}	73	20.0	3.404×10^5
300401-Е	5.410×10^{-6}	4.960×10^{-4}	92	20.0	3.497×10^5

 $< k_2 > (20 \ ^{\circ}\text{C}) = (3.403 \pm 0.049) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$

No	$[E_{1}]_{0}/M$	$[Nuc]_{\circ} / M$	$[Nuc]_{\circ}/[El]_{\circ}$	Conv / %	T / °C	$k_{2} / M^{-1} s^{-1}$
				COIIV. / /0	1 / C	K2 / IVI - S
AO-142.1	3.94×10^{-5}	3.76×10^{-4}	9.5	73	-73.8	5.366×10^2
AO-139.1	$7.90 imes 10^{-5}$	9.66×10^{-4}	12	70	-70.0	6.190×10^{2}
AO-142.2	3.86×10^{-5}	$2.77 imes 10^{-4}$	7.2	58	-60.9	1.075×10^3
AO-142.3	2.33×10^{-5}	1.86×10^{-4}	8.0	65	-50.6	1.830×10^3
AO-142.4	$2.07 imes 10^{-5}$	8.26×10^{-5}	4.0	83	-41.2	2.747×10^3

Table S48. 1-(1-Cyclohexenyl)-4-methylpiperazine (**1q**) and $(dma)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 640$ nm (Schölly).



 $k_2(20 \text{ °C}) = (2.454 \pm 0.188) \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$

Table S49. Pyrrole (4a) and $(pfa)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 600$ nm (Schölly).

No.	[<i>El</i>] ₀ / M	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	Conv. / %	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
210801.PA1	2.616×10^{-5}	9.176×10^{-4}	35	55	20.0	3.083×10^1
210801.PA2	3.242×10^{-5}	1.624×10^{-3}	50	40	20.0	3.313×10^1
210801.PA3	2.839×10^{-5}	3.319×10^{-3}	117	36	20.0	2.973×10^1

 $< k_2 > (20 \ ^{\circ}\text{C}) = (3.123 \pm 0.142) \times 10^1 \text{ M}^{-1} \text{ s}^{-1}$

Table S50. *N*-(Triisopropylsilyl)pyrrole (4c) and $(dpa)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 674$ nm (J&M).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	Conv. / %	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
290801-01	2.596×10^{-5}	1.745×10^{-3}	67	37	20.0	4.367×10^{-2}
290801-02	2.736×10^{-5}	3.218×10^{-3}	118	71	20.0	4.186×10^{-2}
290801-03	2.437×10^{-5}	6.514×10^{-3}	267	93	20.0	4.221×10^{-2}

 $< k_2 > (20 \ ^{\circ}\text{C}) = (4.258 \pm 0.078) \times 10^{-2} \text{ M}^{-1} \text{ s}^{-1}$

Table S51. *N*-(Triisopropylsilyl)pyrrole (**4c**) and $(mfa)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 593$ nm (J&M).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	Conv. / %	T∕°C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
300801-04	1.545×10^{-5}	2.022×10^{-3}	131	70	20.0	1.170×10^{-1}
300801-03	2.035×10^{-5}	3.873×10^{-3}	190	90	20.0	1.271×10^{-1}
300801-05	1.594×10^{-5}	4.172×10^{-3}	262	87	20.0	1.126×10^{-1}

 $< k_2 > (20 \text{ °C}) = (1.189 \pm 0.061) \times 10^{-1} \text{ M}^{-1} \text{ s}^{-1}$

Table S52. *N*-(Triisopropylsilyl)pyrrole (**4c**) and (pfa)₂CH⁺ BF₄⁻ in CH₂Cl₂ at $\lambda = 601$ nm (J&M).

No.	[<i>El</i>] ₀ / M	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	Conv. / %	<i>T</i> / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
300801-02	1.550×10^{-5}	7.682×10^{-4}	50	39	20.0	1.114
300801-01	1.369×10^{-5}	1.695×10^{-3}	124	70	20.0	1.448
300801-03	1.449×10^{-5}	2.872×10^{-3}	198	72	20.0	1.416

 $< k_2 > (20 \text{ °C}) = 1.326 \pm 0.151 \text{ M}^{-1} \text{ s}^{-1}$

Table S53. Indole (6a) and $(pfa)_2CH^+BF_4^-$ in CH_2Cl_2 at $\lambda = 601$ nm (Stopped flow).

_						
	No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
	230501-D	4.997×10^{-6}	$5.250 imes 10^{-4}$	105	20.0	1.298×10^2
	230501-В	4.997×10^{-6}	1.050×10^{-3}	210	20.0	1.367×10^2
	230501-C	4.997×10^{-6}	1.575×10^{-3}	315	20.0	1.352×10^2
1	(1 > (20.90)	(1.220 ± 0.02)	$0 + 10^2 N^{-1}$	-1		

 $< k_2 > (20 \text{ °C}) = (1.339 \pm 0.030) \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$

					1 1
No.	$[El]_0 / M$	$[Nuc]_0 / M$	$[Nuc]_0/[El]_0$	T / °C	$k_2 / M^{-1} s^{-1}$
170501-A	5.297×10^{-6}	9.529×10^{-5}	18	20.0	1.089×10^{3}
170501-Е	5.297×10^{-6}	2.382×10^{-4}	45	20.0	1.074×10^3
170501-F	5.297×10^{-6}	4.764×10^{-4}	90	20.0	1.083×10^3
170501-D	5.297×10^{-6}	7.147×10^{-4}	135	20.0	1.068×10^3
170501-C	5.297×10^{-6}	9.529×10^{-4}	180	20.0	1.110×10^3

Table S54. *N*-Methylindole (**6b**) and $(pfa)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 601$ nm (Stopped flow).

 $< k_2 > (20 \ ^{\circ}C) = (1.085 \pm 0.015) \times 10^3 \ \text{M}^{-1} \ \text{s}^{-1}$

Table S55. 1,2-Dimethylindole (**6c**) and $(pfa)_2CH^+$ BF₄⁻ in CH₂Cl₂ at $\lambda = 601$ nm (Stopped flow).

No.	$[El]_0 / M$	[<i>Nuc</i>] ₀ / M	$[Nuc]_0/[El]_0$	T / °C	$k_2 / \mathrm{M}^{-1} \mathrm{s}^{-1}$
050601-I	4.731×10^{-6}	9.352×10^{-5}	20	20.0	5.666×10^{3}
050601-C	4.731×10^{-6}	1.434×10^{-4}	30	20.0	5.650×10^3
050601-B	4.731×10^{-6}	2.868×10^{-4}	61	20.0	5.219×10^3
050601-A	4.731×10^{-6}	4.301×10^{-4}	91	20.0	5.335×10^3

 $<k_2>(20 \ ^{\circ}C) = (5.468 \pm 0.195) \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$